Letters to the Editor

Synthesis of the first ruthenium-iridium bimetallacarborane cluster

I. A. Lobanova, I. T. Chizhevsky, * P. V. Petrovskii, and V. I. Bregadze

A. N. Nesmeyanov Institute of Organoelement Compounds, Russian Academy of Sciences, 28 ul. Vavilova, 117813 Moscow, Russian Federation.

Fax: +7 (095) 135 5085

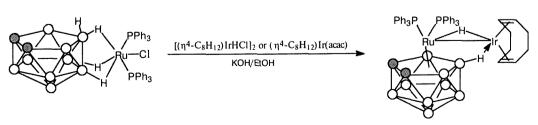
Binuclear metallacarborane clusters containing unusual B-H...M aghostic bonds (where M is a metal) or new B-M-H σ-bonds with the terminal hydride at the metal atom (formed therefrom on intramolecular oxidative addition) attract considerable attention in the chemistry of metallacarboranes. 1,2 In this family of carborane clusters of platinum group metals, compounds incorporating Rh-Rh, 3,4 Rh-Ir, 4 Rh-Ru, 5 and Rh-Pt, 6,7 are already known, and their properties are being intensely studied. In the present paper, we synthesized the first ruthenium-iridium bimetallacarborane $(\eta^4-C_8H_{12})IrRu(\mu-H)(PPh_3)_2(\eta^5-C_2B_9H_{11})$ (1), which is an electron-deficient cluster (32 valence electrons) wherein the bond is formed due to B-H...Ir aghostic interaction and to formation of an Ru-H-Ir bridged system.

We found that the reaction of *exo-nido-* 5,6,10-[Cl(Ph₃P)₂Ru]-5,6,10-(μ -H)₃-10-H-7,8-C₂B₉H₈

(see Ref. 8) with an equimolar amount of dimeric complex $[(\eta^4-C_8H_{12})IrHCl]_2$ (2) in ethanol in the presence of excess KOH at 22 °C followed by recrystallization of the product from a $CH_2Cl_2-C_6H_{14}-n$ mixture affords mixed-metal cluster 1 in 74 % yield. Analytically pure complex 1 was obtained, though in a lower yield (55 %), when starting dimer 2 in the above reaction was replaced by $(\eta^4-C_8H_{12})Ir(acac)$ (Scheme 1).

The composition of cluster 1 was confirmed by the data of elemental analysis. Found (%): C, 50.11; H, 5.11; B, 8.73; P, 5.63. $C_{46}H_{54}B_{9}P_{2}RuIr \cdot 0.5CH_{2}Cl_{2}$. Calculated (%): C, 50.68; H, 5.00; B, 8.83; P, 5.63. The structure of complex 1 is in good agreement with the data of ^{1}H , ^{13}C , ^{31}P , and ^{11}B NMR spectroscopy and with the IR spectra. For example, the ^{1}H NMR spectrum of cluster 1 in $C_{6}D_{6}$ exhibits two high-field signals with a 1: 1 integral intensity ratio. These are a broad quadruplet-like signal at -2.95 ppm ($J_{B-H} \approx$

Scheme 1



80 Hz) and a narrow triplet at -10.70 ppm (J_{H-P} = 19.2 Hz), which were assigned to the protons of B-H...Ir and Ru-H-Ir fragments, respectively. The signals of the H and C atoms of the coordinated diene ligand are exhibited in the ¹H NMR spectrum (in C₆D₆) and ${}^{13}C\{{}^{1}H\}$ NMR spectrum (in CD₂Cl₂) at δ 4.88 (br.s, 4 H, CH=CH), 2.06 (m, 4 H, CH₂), 1.54 (br.q, 4 H, CH₂) and at δ 67.3 (s, CH=CH), 32.0 (s, CH₂); the signals corresponding to the carborane ligand are observed as separate broadened singlets at δ 2.84 and 41.2, respectively. The signals of the phenyl groups of the phosphine ligands are observed in their normal regions: at δ 6.90—7.90 (m, 30 H) in the ¹H NMR spectra and at δ 127.6–137.3 (C_o, C_m, C_p, and C_{kev}) in the ¹³C NMR spectra. The IR spectrum of cluster 1 (pellets with KBr) exhibits a characteristic v(B-H)band at 2580 cm⁻¹ and v(C-H) band at 3070 cm⁻¹; no absorption band corresponding to the bridged hydride was found in the spectrum. The magnetic equivalence of the carbon and hydrogen nuclei in the 1,5-cyclooctadiene and carborane ligands as well as the two equivalent phosphine ligands at the Ru atom [31P{1H} NMR spectrum (C_6D_6) , δ : 50.42 (s)] indicate the presence of symmetry in cluster 1 and suggests that the B(8) atom of the pentagonal open plane of the carborane cage participates in the B-H...Ir aghostic interaction. The typical low-field signal at +6.45 ppm ($J_{B-H} \approx 90$ Hz) in the 11B NMR spectrum of the cluster in CH₂Cl₂ evidences the existence of this interaction; the signals of the other boron atoms of the carborane ligand are observed in a

substantially higher field, namely, in the region from -8.0 to -25.0 ppm.

The work was carried out with financial support of the International Science Foundation (Grant No. M4P 000) and the Russian Foundation for Basic Research (Project No. 93-03-18654).

References

- 1. F. G. A. Stone, Adv. Organomet. Chem., 1990, 31, 53.
- 2. S. A. Brew and F. G. A. Stone, *Adv. Organomet. Chem.*, 1994, **35**, 135.
- P. E. Behnken, T. B. Marder, R. T. Baker, C. B. Knobler, M. R. Thompson, and M. F. Hawthorne, *J. Am. Chem. Soc.*, 1985, 107, 932.
- 4. J. R. Fernandez, G. F. Helm, J. A. K. Howard, M. U. Pilotti, and F. G. A. Stone, *J. Chem. Soc., Dalton Trans.*, 1990, 1747.
- T. Chizhevsky, I. A. Lobanova, I. V. Pisareva, T. V. Zinevich, V. I. Bregadze, A. I. Yanovsky, Yu. T. Struchkov, C. B. Knobler, and M. F. Hawthorne, in *Current Topics in the Chemistry of Boron*, Ed. G. W. Kabalka, Roy. Soc. Chem., Cambridge, 1994, 301.
- J. E. Goldberg, J. A. K. Howard, H. Muller, M. U. Pilotti, and F. G. A. Stone, J. Chem. Soc., Dalton Trans., 1990, 3055.
- J. E. Goldberg, D. E. Mullica, E. L. Sappenfield, and F. G. A. Stone, J. Chem. Soc., Dalton Trans., 1992, 2693.
- 8. I. T. Chizhevsky, I. A. Lobanova, V. I. Bregadze, P. V. Petrovskii, V. A. Antonovich, A. V. Polyakov, A. I. Yanovsky, and Yu. T. Struchkov, *Mendeleev Commun.*, 1991, 47.

Received May 23, 1995

1,3-Dipolar tris-cycloaddition of *tert*-butylphosphaacetylene to 2,4,6-triazido-3-chloro-5-cyanopyridine

S. V. Chapyshev, a* U. Bergstrasser, b and M. Regitzb

^aInstitute of Chemical Physics in Chernogolovka, Russian Academy of Sciences, 142432 Chernogolovka, Moscow Region, Russian Federation.

Fax: +7 (096) 515 3588

^bKaiserslautern University,

Erwin-Schrödinger Strasse, D-67663 Germany

It has been shown recently that 1,3-cycloaddition even of such a reactive dipolarophile as norbornene to 2,4,6-triazido-3-chloro-5-cyanopyridine (1) occurs regioselectively to the azido group at position 4 of the pyridine cycle. 1,2 It has been established that consider-

able weakening of the properties of the α -azido groups in compound 1 as 1,3-dipoles is caused by their strong conjugation with the electron-acceptor pyridine system.³ In this connection, study of the reaction of 1 with tert-butylphosphaacetylene (2) as the dipolarophile with